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Molecular Organization of Comb-Like Polymers Containing Long Fluorocarbon Chains in the Thin Films Formed by the LB and Lift-Up Methods

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Molecular organization in layered structures, which can be caused by side-chain crystallization of a comb-like polymer having long fluorocarbon chains, and the structure of thin films formed by LB and lift-up methods, have been investigated, in comparison with those of poly(octadecylacrylate).

Keywords: molecular organization; comb-like polymers; fluorocarbon chains; thin film; LB method; lift-up method

INTRODUCTION

A comb-like polymer containing a sufficiently long fluorocarbon chain in the side chains is crystallized easily. Because fluorine atom is too large to allow a precise transzig-zag planar conformation of $-(CF_2-CF_2)_n$ — chain, the side chains are rather rigid and the cohesive force among them is strong. In this paper, molecular organization in the thin films formed by the LB and lift-up methods has been examined on the bases of cohesive energy and conformational freedom of the comb-like polymer chains.

EXPERIMENTAL

IH, IH, IIH-icosafluoroundecyl acrylate $(H(CF_2)_{\pm 0}CH_2OCOCH=CH_2, F_{\pm 0}A)$, mp 44.0-45.4°C); IH, IH, 9H-hexadecafluorononyl acrylate $(H(CF_2)_8CH_2OCOCH=CH_2, F_8A)$, mp -21.6°C) were bought from Daikin

Fine Chemicals. $F_{1,0}A$ was used after purification by recrystallization from ethanol solution, while F_0A was used as received. Their polymers $(PF_{1,0}A)$ and PF_0A and poly(octadecyl acrylate) $(PC_{1,0}A)$ were prepared by γ -ray-irradiation post-polymerization. Thermal properties of these polymers were measured using a Seiko Instruments Model DSC6200 differential scanning calorimeter (DSC). For additional characterization of the film structure, a JEOL FT-IR spectrophotometer Model JIR-100 and a Rigaku diffractometer RAD-B with Cu K4 radiation were used.

Monolayer of $PF_{1,0}A$ was spread from methylethylketone solution $(6.3\times10^{-4}\,\mathrm{M})$ and surface pressure-area $(\pi$ -A) isotherms were measured using a Lauda film balance. LB films were deposited on a CaF_2 plate using a KSV apparatus (LB2200) with a dipping speed of 10 cm min⁻¹ at 30 mN m⁻¹. In the case of lift-up method, the CaF_2 plate of hydrophobic surface with a ferric stearate monolayer immersed vertically into the dilute solution of $PF_{1,0}A$, and then it was slowly withdrawn.

RESULTS AND DISCUSSION

Estimation of Cohesive Energy Among Side Chains

Cohesive energies among fluorocarbon chains in the comblike polymers were estimated by the DSC measurements. Thermal property of PF_{1D}A,PF₈A and PC₁₈A are summarized in Table 1, where T_m and T_c are temperatures of melting and crystallization, ΔH_{\perp} and ΔH_{c} are molar heats of fusion and crystallization per repeating unit, respectively, and ΔS_{\perp} and ΔS_{c} are corresponding entropy changes.

Table	1 Thormal	Properties	a f	Comb 14	ka Da	Lymore
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Sample	τ	τ	ΔH	ΔH_c	ΔS_{i}	ΔS	
Sample	4 m 4 c		kJmol ⁻¹		JK- mol-		
PF _{1.0} A	118.2	105.1	12.6	11.4	32.2	30.1	
PF ₈ A	54.7	19.9	6.1	3.7	18.5	12.6	
$PC_{18}A$	53.7	46.3	22.7	19.3	69.5	60.2	

The values of ΔH_c seem to reflect the cohesive energies and those of ΔS_c indicate regularity of lammelar structure formed by side chains. Values of ΔH_c and ΔS_c for $PF_{10}A$ are about half of those for $PC_{18}A$. Thus, it can be expected that $PF_{10}A$ has a sufficient cohesive energy among the side chains to form the thin solid film with lamellar structure, whereas cohesive energy for $PF_{8}A$ is insufficient.

Film Formation by LB and Lift-up Methods

These comb-like plymers can form stable condensed monolayers at the air-water interface up to 70 mNm⁻¹. From the π -A isotherms, as shown in Fig.1, it has been found that the monolayer of PF_{1.0}A has a tendency to condense with the increase of temperature, whereas that of PC_{1.8}A tends to expand. The limiting areas per side chain of PF_{1.0}A were 0.23 nm⁻² at 15°C and 0.21 nm⁻² at 25°C, while those of PC_{1.8}A were 0.27 nm⁻² at 15°C and 0.29 nm⁻² at 25°C. Therefore, formation of LB films at a higher temperature seems to be rather preferable for PF_{1.0}A, while a lower temperature will be better for PC_{1.8}A.

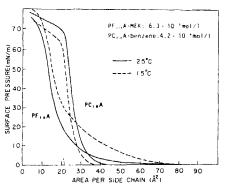


FIGURE 1 π -A isotherms of PF₁₀A and PC₁₈A.

In the case of hydrohobic plate, only first cycle was Y-type deposition and followed by X-type deposition. On the contrary, with the hydrophilic plate, first two cycles were Z-type deposition and further deposition was difficult.

The thin film of PF₁₀A formed by the lift-up method has a lamellar structure which is similar to the LB film. This can be ascribed to the adsorption of the comb-like polymers on the solid surface followed by the side-chain crystallization. Film thickness, as estimated by IR measurements, was decreased with the increasing of lift-up speed, and in the case of speed faster than 0.10 cm min⁻¹ the film formation by this method was impossible. The adsorption of comb-like polymer as well as the side-chain crystallization is considered to be a time-consuming process because of the entanglement of comb-like polymer chains. Film thickness increased linearly with the solution concentration.

X-Ray diffraction patterns, as shown in Fig.2, indicate that regularity of the lamellar structure of the PF_{1.0}A films formed by the lift-up method is less than those in the LB film but better than those in cast film.

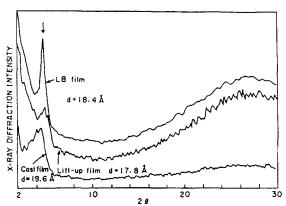


FIGURE 2 X-Ray diffraction patterns for films of PF10A.

In these cases, the structure of films is considered as a single layer rather than an interdigitated anti-parallel packing of side chains [4].

Mechanism of film formation on the solid surface by the lift-up method can be considered as follows. When the solid plate is immersed in the solution of $PF_{10}A$, at first the comblike polymers are adsorbed on the surface, followed by self-organization of long fluorocarbon chains. Additional polymer chains may be deposited successively on the initially adsorbed polymer, and rearranged forming a somewhat ordered lamellar structure. Furthermore, in the lift-up process of solid plate, according to evaporation of solvent the local concentration of $PF_{10}A$ at meniscus on the plate should be increased.

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